

Unit 1 Study Guide --- ANSWER KEY

1. Define scientific method: a procedure for collecting, testing, and communicating information
2. Define hypothesis: a suggested answer to the question, an educated guess
3. Define experiment: a scientific test to see what will happen
4. Define dependent variable: the variable that you do NOT control. (on a graph is usually on the Y axis)
5. Define independent variable: the variable that I control. (on a graph is usually on the X axis)
6. Define control: the part of the experiment where everything is left the same.
7. Define observation: information collected with any of your 5 senses
8. Define density: the ratio of mass per unit volume ($D = m/v$)
9. Define accuracy: the measure of how close your data is to the ACTUAL or true value.
10. Define precision: a measure of how close your data (measurements) are to each other.
11. Why do scientists use SI units of measurement? So that measurements can be understood internationally (around the world).

12. List (in the correct order) the steps of the scientific method.

1. make an observation	4. test hypothesis (experiment)
2. ask a question	5. analyze data/form a conclusion
3. form a hypothesis	6. develop a theory

13. What mnemonic device is used to remember the metric system?

K H Da B d c m
g, m, L

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	K	H	Da	g, m, L	d	c	m
Meaning in powers of 10	10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}

Convert the following:

16. 25L to 25000 mL
17. 2.3 km to 230,000 cm
18. 0.045 g to 45 mg
19. 0.35 kg to 350 g
20. 300 mL to 0.300 L

21. 25 cm to 0.25 m
22. 56784 mm to 0.056784 km
23. 15 g to 0.015 kg
24. 3.452 kL to 3452 L
25. 47 m to 47000 mm

Scientific Notation:

Convert the following to scientific notation:

- 26) 0.005 5×10^{-3}
- 27) 5,050 5.05×10^3
- 28) 0.0008 8×10^{-4}
- 29) 1,000 1×10^3

- 30) 0.025 2.5×10^{-2}
- 31) 0.0025 2.5×10^{-3}
- 32) 500 5×10^2
- 33) 5,000 5×10^3

Carbon Compound Study Guide Answer Key

**Katherine J. Denniston, Denniston
Katherine**

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Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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Chemistry, Student Study Guide John A. Olmsted,Gregory M. Williams,2002-01-04 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and analytical techniques have been updated there is now an additional third chapter on chemical equilibrium The authors have also developed an expanded and more integrated problem solving emphasis that now incorporates a 4 step strategy throughout complete with text icons The whole is backed by a range of supplements including a new illustration program a tutorial CD interactive learningware an extensive Web CT component an instructor s resource CD and a solution CD

Study Guide and Full Solutions Manual Susan McMurry,John McMurry,2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook

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Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Student Study Guide to Accompany Fox's Physiological Basis for Exercise and Sport** Susan Muller,Steven J. Keteyian,Merle L. Foss,1998 Written with the physical educator exercise specialist physical therapist and coach in mind this new edition continues to offer the clear presentation insightful explanations depth of coverage and straightforward writing style that students and instructors have come to expect from this leading text originally authored by Edward Fox Inside you'll find the strongest presentation of aerobic and anaerobic training programs and responses available **Organic Chemistry, Study Guide and Solutions Manual** T. W. Graham Solomons,Craig B. Fryhle,2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest The bark of the Pacific yew tree produces Taxol found to be a highly effective drug against ovarian and breast cancer Taxol blocks mitosis during eukaryotic cell division The supply of Taxol from the Pacific yew tree is vanishingly small however A single 100 year old tree provides only about one dose of the drug roughly 300 mg For this reason as well as the spectacular molecular architecture of Taxol synthetic organic chemists fiercely undertook efforts to synthesize it Five total syntheses of Taxol have thus far been reported Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand This case clearly demonstrates the importance of synthesis and the use of organic chemistry It's just one of the many examples used in the text that will spark the interest of students and get them involved in the study of organic chemistry **Chemistry** Stanley R. Radel,Marjorie H. Navidi,1990

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