

## 20.12 Cyclohexane: a cyclic hydrocarbon

### Physical properties and structure

Cyclohexane is a colourless, odourless liquid with a boiling point of 98 °C and a melting point of 6 °C. It is insoluble in water but soluble in organic solvents.

The structure of cyclohexane is a hexagonal ring of six carbon atoms, each bonded to two hydrogen atoms. The ring is not planar but adopts a chair conformation to minimize steric strain.

The chair conformation of cyclohexane is the most stable arrangement of the ring, with all carbon-carbon bonds in a staggered conformation.

The chair conformation of cyclohexane is shown in the diagram below. The axial bonds are perpendicular to the ring plane, while the equatorial bonds are in the plane of the ring.

### Chemical properties and reactions

Cyclohexane is a saturated hydrocarbon and is relatively inert to most chemical reactions. It can undergo combustion to form carbon dioxide and water.

### Applications

Cyclohexane is used in the production of cyclohexanone and cyclohexanone oxime, which are important intermediates in the synthesis of various chemicals.

Cyclohexane is also used as a solvent in the extraction of natural products and in the synthesis of polymers.

### Structure and properties of cyclohexane

Cyclohexane is a saturated hydrocarbon with the molecular formula  $C_6H_{12}$ . It is a colourless, odourless liquid with a boiling point of 98 °C and a melting point of 6 °C.

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